

Chapter 10 Chemical Quantities Vocabulary Review Answers



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Chapter 10 "Chemical Quantities" Vocab. Tools. Copy this to my account; E-mail to a friend; Find other activities; Start over; Help; Use these activities to learn the vocabulary presented in this chapter. A B; percent composition: a description of the relative amounts of each element in a compound ... the SI unit representing 6.02×10^{23} ...

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substance.Chapter 10 - Chemical Quantities CHAPTER 10: Chemical Quantities BASICS: • The basic unit that is used to determine the amount of a chemical substance is called a mole • A mole(mol) of a substance is equivalent to 6.02×10^{23} particles of that substance • The mole was founded by a

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Chapter 10 Chemical Quantities Worksheet Chapter 10 chemical quantities, chapter 10 chemical quantities jennie l borders section 101 the mole: a measurement of matter you often measure the amount of something by count, by mass, or by volume a mole (mol) of a substance is $6.02 \times$

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